Lecture Notes 11

Bose-Einstein Condensation

In Lecture 10 we obtained the following formulas for N and U:

$$N = \frac{gV}{\lambda_T^3} F_{3/2}(\xi) \tag{1}$$

$$U = \frac{gV}{\lambda_T^{3}} \frac{3}{2} k_B T F_{5/2}(\xi)$$
(2)

The thermal (de Broglie) wavelength is: $\lambda_T = \frac{h}{\sqrt{2\pi m k_B T}}$ and

$$F_m(\xi) = \frac{1}{\Gamma(m)} \int_0^\infty \frac{x^{m-1}}{e^{x\xi^{-1} - 1}} dx$$
(3)

Since U = (3/2)pV we get from Eq.(2):

$$p = \frac{gk_BT}{\lambda_T^3} F_{5/2}(\xi) \tag{4}$$

Einstein observed in 1925 that if the number of particles is conserved, the ideal gas will undergo a phase transition at low enough temperatures. We fix the number of bosons per volume N/gV, while lowering temperature and thus increasing the thermal wavelength. In Figure 1 we show $N\lambda_T^3/gV$ vs fugacity ξ :



Figure 1: $(N/gV)\lambda_T^3 = F(3/2, \xi) vs \xi$

For $N\lambda_T^3/gV < \zeta(3/2) = 2.612$ the fugacity is $\xi < 1$. For $N\lambda_T^3/gV > \zeta(3/2)$ the fugacity is equal to unity $\xi = 1$, or the chemical potential equals zero $\mu = 0$. At these low temperatures the gas condenses in the ground state: n_0/V is finite. Equation 1 is modified to fix this breakdown of the integral approximation:

$$\frac{N-n_0}{V} = \frac{g}{\lambda_T^3} F_{3/2}(1)$$
(5)

At the condensation temperature:

$$\frac{N}{V} = \frac{g}{\lambda_{TC}^{3}} F_{3/2}(1) = \frac{g}{\lambda_{TC}^{3}} \zeta(3/2)$$
(6)

Divide Eq. (5) by Eq. (6):

$$\frac{N-n_0}{N} = \frac{\lambda_{TC}^3}{\lambda_T^3} \tag{7}$$

Using the definition for thermal wavelength:

$$\frac{n_0}{N} = 1 - \left(\frac{T}{T_c}\right)^{3/2} \tag{8}$$

Hence at T = 0K: $n_0/N = 1$ because all bosons are in the ground state. At $T = T_C$: $n_0/N = 0$ as the fraction of all bosons in any state is negligible.

The phase diagram in the temperature, pressure plane is obtained by substituting $\xi = 1$ and $T = T_C$ in Eq. (4):

$$p = \frac{gk_B T_C}{\lambda_{TC}^3} F_{5/2}(1) = \frac{gk_B T_C}{\lambda_{TC}^3} \zeta(5/2)$$
(9)

where $\zeta(5/2) = 1.341$.

In the condensate, $T \le T_C$, the energy is obtained by substituting $\xi = 1$ in Equation (2):

$$U = \frac{gV}{\lambda_T^3 2} k_B T F_{5/2}(1) = \frac{gV}{\lambda_T^3 2} k_B T \zeta(5/2)$$
(10)

We calculate energy per boson by dividing Eq. (10) by Eq. (6):

$$\frac{U}{N} = \frac{3}{2} k_B T \frac{\zeta(\frac{5}{2})}{\zeta(\frac{3}{2})} (\frac{T}{T_c})^{3/2}$$
(11)

The isochoric heat capacity is obtained by differentiating Eq. (11) at fixed V and N:

$$C_V = \frac{\partial U}{\partial T} = 1.925 N k_B \left(\frac{T}{T_c}\right)^{3/2}$$
(12)

Note that at the condensation temperature $C_V = 1.925 Nk_B > 1.5 Nk_B$, the high temperature, classical, heat capacity.

Appendix

We study the special functions $F_m(\xi)$ introduced in Eq.(12).

$$F_m(\xi) = \frac{1}{\Gamma(m)} \int_0^\infty \frac{x^{m-1}}{e^x \xi^{-1} - 1} dx$$
(23)

We can rewrite Eq (23) as:

$$F_m(\xi) = \sum_{n=1}^{\infty} \frac{\xi^n}{n^m}$$
(26)

In the classical limit, $\xi \ll 1$, and $F_m(\xi) \simeq \xi$, as we have already shown to be the case in Eq.(19). In the quantum limit, where Bose-Einstein condensation occurs, $\xi = 1$:

$$F_m(1) = \sum_{n=1}^{\infty} \frac{1}{n^m} = \zeta(m)$$
(27)

Here we introduced the Riemann zeta function $\zeta(m)$. The zeta function of real variable was used first by Euler. Here are a couple of values that are used in the study of Bose condensation: $\zeta(3/2) = 2.612375$, $\zeta(5/2) = 1.341488$. The zeta function $\zeta(m)$ is finite for m > 1. It diverges as m approaches 1:



Figure 2: Riemann zeta function $\zeta(m)$ *diverges as m approaches unity from above.*

<u>Readings</u> Callen Ch. 18 Problem Set 10 Computer lab 6